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Description

In one aspect, this invention relates to an improved hair spray resin, particularly adapted for use in aerosol and pump formulations, having excellent hair holding capacity under conditions of high humidity and improved compatibility with alcohols and hydrocarbon propellants. In another aspect, the invention relates to a suspension polymerization for the preparation of microspheric particles of said improved resins in high yields.

Generally, effective hair spray formulations and film forming resins must satisfy a rigid set of requirements. Specifically, the resulting films should remain non-tacky in humid atmosphere but should be easily removable in aqueous soap solutions or shampoos. The films should have high hair cohesivity and yet possess sufficient strength and elasticity so as to avoid dusting or flaking when the hair is subjected to varying stresses during combing or brushing. The resulting films should remain clear, transparent and glossy on ageing and are required to have a viscosity range which permits uniform spraying without nozzle clogging. Additionally, the hair spray resin should show little or no tendency to interact with perfume or other optional components conventionally utilized in hair spray formulations and should be readily soluble in organic solvents while exhibiting good compatibility with conventional hydrocarbon propellants.

Many polymeric systems have been utilized in an attempt to meet these stringent requirements. Among these are included polyvinylpyrrolidone and copolymers of N-vinylpyrrolidone with vinyl acetate. However, these copolymers do not exhibit the desired degree of holding at high humidity. Moreover, several of the vinylpyrrolidone polymers possess an unpleasant odor. Vinyl ether/maleic half ester copolymers have also been used in hair sprays, but high molecular weight is required to achieve adequate holding. Additionally, vinyl acetate polymers having 15 to 35 % of their acetate groups converted to hydroxyl groups have been proposed for increasing solubility in carbon dioxide propellant systems. Unfortunately, such increased solubility is achieved at the cost of lowered holding power. Although each of the above resins meets at least some of the above cited requirements, none has exhibited all of these characteristics to a desired degree.

Accordingly, it is an object of the present invention to overcome the individual deficiencies of the above discussed polymers and to provide an improved alcohol soluble resin ideally suited for hair sprays in a particulate microspherical form.

It is another object of this invention to provide a convenient and economical process for the preparation of said resin.

Still another object of this invention is to provide a microspheric resin for use in hair sprays and having excellent hair holding capability, particularly under conditions of high humidity.

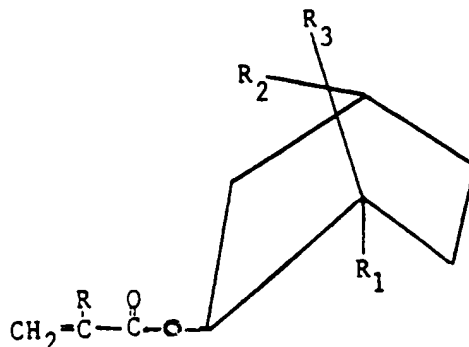
Another object is to provide an esteric resin formulation which eliminates spray can corrosion.

Another object is to provide a relatively high molecular weight esteric resin having improved propellant compatibility, excellent spray pattern, increased hair holding and a superior low cloud/clear temperature point.

These and other objects will become apparent from the following description and disclosure.

According to this invention there is provided a terpolymer resin of random or alternating structure comprising a vinyl ester, a water insoluble or water miscible alkyl maleate half ester and the acrylate or methacrylate ester of a saturated hydroxylated bicyclic hydrocarbon in a molar ratio of 1:0.35-1:0.05-0.25.

The proportions of monomers in the present resin are critical since it is found that, above 3.5:1 vinyl acetate to maleate half ester, the terpolymer becomes insoluble in commercial alcohol carriers such as ethanol and below 1:1 the terpolymer develops tack. It is also found that the high water sensitivity and/or low glass transition temperatures (T_g) of the vinyl ester and maleate half ester copolymers prevents their synthesis by the economical one step suspension polymerization of the present invention. Accordingly, the incorporation of an acrylate or methacrylate ester of a saturated, hydroxylated bicyclic hydrocarbon having a relatively high T_g , is an essential monomer in the terpolymeric compound. Acrylate and/or methacrylate esters of isoborneol, exo-norborneol and endo-norborneol are preferred and the isobornyl ester is the most preferred. The preferred bicyclic compounds are those having a hydrocarbon bridge and are generally defined by the formula



wherein each of R, R₁, R₂ and R₃ is hydrogen or methyl. These monomers minimize development of tack in the hair spray formulation. The isobornyl acrylate and isobornyl methacrylate, because of their high homopolymer T_g values, i.e. 90–100°C. and 170°C., respectively, are particularly suitable for this purpose. In general, a T_g of from about 40°C. to about 100°C. is desirable in the present terpolymeric product.

According to this invention, the molar ratio of vinyl ester to the bicyclic acrylate or methacrylate monomer in the terpolymer should be between about 4:1 and about 20:1. A particularly preferred molar ratio of vinyl acetate/mono-*n*-C₄ to C₅ alkyl maleate/isobornyl acrylate or methacrylate is about 1:0.6–0.8:0.08–0.12, e.g. 1:0.75:0.1.

Illustrative of the water insoluble or water miscible alkyl maleate half ester monomers of this invention are the propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl and decyl half esters; although the *n*-butyl and *n*-pentyl half esters are the most preferred.

Examples of the vinyl ester monomeric component are those containing 4 to 14 carbon atoms which include vinyl acetate, vinyl propionate, vinyl isopropionate, vinyl isobutyrate, vinyl butyrate, vinyl hexanoate, vinyl pivalate, vinyl laurate and vinyl neodecanoate, of which vinyl acetate is the most preferred.

The present resin in the form of beads or microspheres prepared by suspension polymerization provides certain advantages. For example, the resin has a higher molecular weight for better hair holding under humid conditions and achieves the highest conversion, thus minimizing unreacted monomers. Additionally, the present resin in dry bead form can be handled and shipped safely in the absence of flammable alcohol solvents.

The microspheres of the present terpolymeric resin generally have an average diameter of between about 0.5 and about 2.5 mm when synthesized by the suspension polymerization process of the present invention and the polymer products generally possess a relative viscosity (RV) greater than 1.3. Below a relative viscosity of 1.3 the polymer lacks curl retention under humid conditions.

Generally, the suspension polymerization process comprises adding the monomers, individually or pre-mixed in the above proportions, to a 0.1 to 2.0 % by weight aqueous suspension medium, preferably a 0.25 to 1.0 weight % solution, of a carboxylated polyelectrolyte, preferably methyl vinyl ether–maleic acid copolymer (GANTREZ® S-95), wherein they are polymerized in the presence of a free radical initiator under conditions of agitation at a temperature of between 40° and 90°C., preferably between 50° and 70°C. The initiator can be added to the monomer mixture before or after the monomers have been charged into the suspension medium. The reaction is executed in an inert atmosphere which can be maintained by purging with nitrogen to eliminate air and oxygen.

The polymerization reaction is carried out under constant agitation over a period of from about 4 to about 12 hours; although usually 6 to 8 hours is sufficient to complete the reaction and formation of a bead-like terpolymer product. The beads are then separated from the suspension medium, washed with water, and dried. The supernate separated from the product can be recycled to the reaction zone if desired as a make-up suspension media in which any unreacted monomer in the supernate can be converted, thus providing a highly efficient, pollution-free process.

Generally, the suspension medium can be any aqueous solution containing from 0.1 wt. % to 2.0 wt. % of a carboxylated polyelectrolyte suspending agent. Suitable suspending agents include methyl vinyl ether/maleic acid copolymer (e.g. GANTREZ S-95)*; ethylen /maleic anhydrid copolymer (e.g. EMA

* Supplied by GAF Corporation

resin[™]; vinyl acetate/maleic anhydride copolymer and N-vinyl-2-pyrrolidone/maleic anhydride copolymer[™]. The concentration of total monomers in the suspension medium can vary from 10 wt. % to 50 wt. %; however, concentrations of between about 20 and about 40 wt. % monomers are recommended from the standpoint of economics and process conditions.

5 Generally, between 0.05 wt. % and 5.0 wt. % of initiator, based on total monomers of the terpolymer can be employed in the polymerization reaction; although, in most instances, between about 0.1 and about 2.0 wt. % initiator is sufficient to promote the reaction.

The free radical initiators employed in the present reaction are typically low temperature initiators having a half-life of 10 hours at temperatures between about 45°C. and about 65°C.; although any of the free radical initiators which are effective at temperatures between about 40°C. and about 80°C. are suitably employed herein. Typical low temperature initiators include peroxyesters such as t-butylperoxypivalate (Lupersol® 11) and t-butylperoxyneodecanoate (Lupersol® 10), peroxydicarbonates such as di-(n-propyl)peroxydicarbonate (Lupersol® 221), di-(sec-butyl)peroxydicarbonate (Lupersol® 225) and di-(2-ethylhexyl)peroxydicarbonate (Lupersol® 223) all supplied by Penwalt Corporation. Also, azo initiators such as 2,2'-azobis(2,4-dimethylvaleronitrile), VAZO®-52 and 2,2'-azobis(2,4-dimethyl-4-methoxy-valeronitrile), VAZO®-33W, both supplied by duPont, are suitable.

The present terpolymeric resins are employed as the active ingredients in any of the standard hair treating formulations employed for pump and aerosol sprays. The resins can also be used to augment existing hair spray formulations to improve solubility, hair holding and hydrocarbon compatibility. When used as the sole active hair holding agent in the formulation, the present resins are employed in concentrations between 0.5% and 10% solids, preferably between about 2.0% and about 5.0% solids.

In preparing the hair treatment formulations, the resin is usually dissolved in an inert carrier, such as a lower alcohol, e.g. ethanol, an aqueous ethanol solution, isopropanol and, the like, etc. For aerosol sprays, the formulations also include a conventional propellant such as, for example, a 20/80 blend of propane/isobutane (Propellant A-46), nitrogen, nitrogen oxide, carbon dioxide, etc.

25 The formulations are charged into a canister and the propellant pressurized into the canister to provide a spray operated through a pressure release nozzle. The present resins have a long shelf life and avoid nozzle clogging or canister corrosion when employed in the above concentrations.

Having thus generally described the invention, reference is now had to the following examples which provide specific and preferred embodiments but which are not to be construed as limiting to the scope of the invention as more broadly described above and in the appended claims. All parts given are by weight unless otherwise indicated.

EXAMPLE 1

35 Into a four-necked, one-liter resin kettle, fitted with a mechanical agitator, a reflux condenser, a dropping funnel, thermometer and a nitrogen inlet tube was charged:

86.00 g. of vinyl acetate (1.00 mole)
103.20 g. of mono-n-butyl maleate (0.06 mole)
40 20.80 g. of isobornyl acrylate (0.10 mole) and
4.20 g. of t-butylperoxyneodecanoate
(Lupersol® - 10 M 75; 75% active)

The contents of the kettle were mixed thoroughly by the mechanical agitator operating at about 250 rpm for a period of about 5 minutes, after which a polyelectrolyte solution prepared from 1.05 g. of GANTREZ® S-95 and 315.00 g. of distilled water was added.

45 The mixture was purged with nitrogen and gently heated to 62°C. over a period of 60 minutes while maintaining agitation and the nitrogen purge. The reactants were held at 62-64°C. for 8 hours during which the bead-like resin product was formed. The mixture was then cooled to 15°C. in a water/ice bath and the supernate decanted. The remaining beads, having an average diameter of 0.5-1.0 mm, were washed twice with distilled water, filtered and then dried under house vacuum at ambient temperature for 8 hours, followed by drying in a hot air oven at 55-60°C. for 24 hours.

The supernate separated from the crude product can be recycled to the reaction zone to provide a portion of the suspension medium, if desired.

55 The terpolymeric product, referred to as Polymer 1, in the form of microspheric particles and having the composition:
40.95% vinyl acetate (VAC)

** Supplied by Monsanto Co.

49.14% mono-n-butyl maleate (MBM) and

9.91% isobornyl acrylate (IBA)

was recovered in 87.8% yield. This polymer has a relative viscosity of 1.66, a K-value of 49.6, a molecular weight of 370,000 (by gel permeation chromatography in tetrahydrofuran), an acid number of 129.4 mg.

5 NaOH/g and a Tg of 70°C.

EXAMPLES 2-10

10 In Examples 2-10 the procedure identical to that described in Example 1, with the below noted modifications, was repeated to prepare Polymers 2-10.

Examples 2-7 were modified in that the weight ratios of the monomers and the polymerization initiators were altered as shown in Table I. Also, in Example 4 the GANTREZ® S-95 suspending agent was replaced by ethylene/maleic anhydride copolymer (EMA® resin). In Example 8, the isobornyl acrylate monomer and the t-butylperoxy neodecanoate initiator were replaced by isobornyl methacrylate and di-
15 (2-ethylhexyl) peroxydicarbonate (Lupersol® 223 M75), respectively, and the weight ratios of the monomers were also varied as noted in Table I. In Example 9, the mono-n-butyl maleate monomer and the t-butylperoxyneodecanoate initiator were replaced by mono-n-pentyl maleate and di-(2-ethylhexyl) peroxydicarbonate (Lupersol® 223 M75), respectively, and the weight ratios of the monomers were also
20 varied as noted. In Example 10, t-butylperoxyneodecanoate was replaced by Lupersol® 223 M75, mono-n-butyl maleate was replaced by a mixture of mono-n-butyl maleate and mono-n-pentyl maleate and the weight ratios of the monomers were also altered as noted.

Each of the polymeric products, i.e. Polymers 1-10, were analyzed and characterized as reported in following Table I.

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TABLE I

Polymer	1	2	3	4	5	6	7	8	9	10
Vinyl Acetate (VAC)	40.95	37.26	32.43	33.83	27.74	32.68	36.47	36.26	34.92	35.68
Mono-n-butyl Maleate (MBM)	49.14	44.71	51.89	57.99	55.48	56.03	54.71	54.38		26.76
Mono-n-pentyl Maleate (MPM)									56.64	28.94
Isobornyl Acrylate (IBA)	9.91	18.03	15.68	8.18	16.78	11.29	8.82		8.44	8.62
Isobornyl Methacrylate (IBMA)								9.36		
t-Butylperoxy neodecanoate (Lupersol® 10M75), %	1.50									
t-Butylperoxy pivalate (Lupersol® 11), %	1.00									
Di-(2-ethylhexyl) peroxydicarbonate (Lupersol® 223M75)			1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
GANTREZ® S-95 in water, %	0.333	0.500	0.333		0.333	0.333	0.250	0.333	0.333	0.333
EMA® Resin in water, 1%				0.333						
Relative Viscosity	1.66	1.73	1.60	1.72	1.41	1.53	1.64	1.34	1.61	1.63
K Value	49.6	51.8	47.3	51.3	39.5	44.6	48.5	35.6	47.8	48.4
Molecular Weight $\times 10^{-3}$	370	430	480	440	380	260	500	120	570	550
Acid Number	129.4	120.0	124.3	143.5	142.8	131.0	135.4	138.7	123.3	132.3
Tg, °C.	70	73	71	69	51	72	79	79	72	86

EXAMPLE 11

This example illustrates the excellent solution properties of the hair spray resins of this invention.

Each resin of Examples 1-10 were evaluated for its solubility in ethanol without adding a neutralizing agent such as 2-amino-2-methyl-1-propanol (AMP). In these tests, 2.5 g. of each polymer were

dissolved in 67.5 g. of anhydrous ethanol (SDA 40-2 grade) under agitation and the clarity of the solution was noted. For each clear solution, 30 g. of n-hexane were then added and the solubility in ethanol/n-hexane mixture was recorded. In this test, n-hexane was found to be an ideal substitute for evaluating the compatibility of resin with hydrocarbon propellant A-46 (blend of 20/80 propane/isobutane). The testing results were compared with Resyn® 28-2930* (available from National Starch and Chemical corporation) as shown in following Table II.

TABLE II

Polymer	Solubility in Ethanol	Solubility in Ethanol/n - Hexane
Resyn® 28-2930 (control)	slightly hazy	hazy
1	clear	clear
2	clear	clear
3	clear	clear
4	clear	clear
5	clear	clear
6	clear	clear
7	clear	clear
8	clear	clear
9	clear	clear
10	clear	clear

In addition, all of the polymers of this invention (Polymers 1-10) in the ethanol/n-hexane mixture remained clear at a temperature below -18°C. (0°F.) which is an indication of their excellent compatibility with hydrocarbon propellant for aerosol hair spray applications.

EXAMPLE 12

This example provides a representative formulation for an aerosol hair spray employing a typical hydrocarbon C₃-C₄ propellant. The resins were each dissolved in anhydrous ethanol, 100% neutralized with 2-amino-2-methyl-1-propanol (AMP) and charged to an aerosol spray can with a hydrocarbon propellant A-46 (blend of 80/20 propane/isobutene), as noted in following Table III.

TABLE III

	Formulation A		Formulation B	
	Wt., g.	Wt. %	Wt., g.	Wt. %
Resyn® 28-2930 (control)	2.5	2.5	-	-
Hair spray resins of this invention	-	-	2.5	2.5
Anhydrous ethanol (SDA 40-2)	67.5	67.5	67.5	67.5
Propellant A-46	30.0	30.0	30.0	30.0
2-Amino-2-methyl-1-propanol (AMP)	"	"	"	"
* As required to 100% neutralization of the carboxyl functionalities of the hair spray resin.				

The aerosol hair sprays thus obtained were evaluated for their hair holding properties. The clean, dried tresses were each sprayed with the above formulations 3 seconds at a distance of 50.8 mm (2 inches), combed through twice, rolled on 15.9 mm (5/8") roller and dried under a salon type drier for one hour. These tresses were then unrolled on a humidity rack and placed in a 26.7°C (80°F), 90% relative humidity cabinet for 90 minutes. The curl readings at 60 and 90 minutes were recorded. The results of the relative curl retention at 60 and 90 minutes intervals, using Resyn® 28-2930 as the control which is assigned as a value of 1.00, are shown in the following TABLE IV.

* Vinyl acetate/crotonic acid/vinyl neodecanoate terpolymer, a commercial hair spray resin from the National Starch and Chemical Corporation.

TABLE IV

Hair Spray Resin	Curl Retention Relative to Resyn 28-2930 ^{***}	
	60 minutes	90 minutes
Resyn® 28-2930 (control)	1.00	1.00
Polymer 1	1.09	1.09
Polymer 2	1.09	1.12
Polymer 3	1.30	1.16
Polymer 4	1.09	1.10
Polymer 5	1.01	1.03
Polymer 6	1.06	1.03
Polymer 7	1.23	1.13
Polymer 8	1.05	1.02
Polymer 9	1.14	1.16
Polymer 10	1.14	1.13

^{***} Differences of 0.10 or greater represent significant differences in hair holding performance.

The above data show that the vinyl acetate/mono-n-alkyl maleate/isobornyl acrylate and vinyl acetate/mono-n-alkyl maleate/isobornyl methacrylate polymers of this invention are significantly superior in high humidity holding (60 and 90 minute intervals at 26.7°C (80°F.) and 90% relative humidity) to a commercial hair spray resin such as Resyn® 28-2930.

After curl retention tests were completed, each of the polymer films was easily removed in aqueous soap solutions or shampoos, leaving no detectable residues.

EXAMPLE 13 (COMPARATIVE)

As taught in the art by Example 7 of U.S. Patent 4,567,035, equimolar amounts of vinyl acetate and monoethyl maleate in a solution (ethanol) polymerization process failed to form the desired vinyl acetate/monoethyl maleate alternating copolymer for use as the hair spray resin. The following example was carried out to demonstrate that the suspension polymerization of vinyl acetate and mono-n-butyl maleate in water also failed to produce the discrete, microspheric resins suitable for use as the hair spray resins. Mono-n-butyl maleate was substituted for monoethyl maleate since the latter is soluble in water and is not suitable for the suspension polymerization method.

A vinyl acetate/mono-n-butyl maleate copolymer (Polymer 11) with a molar ratio of vinyl acetate (0.8 mole) and mono-n-butyl maleate (0.6 mole), identical to the Polymer 7, was prepared under the polymerization conditions of foregoing Example 7, except no bicyclic ester, isobornyl acrylate or isobornyl methacrylate, or a mixture thereof was incorporated. The resultant polymer was a sticky, elastic mass containing unreacted monomers and thus was not suitable for hair spray applications.

This example establishes that the incorporation of the present bicyclic ester monomer, e.g. isobornyl acrylate, exo-norbornyl acrylate, endo-norbornyl acrylate, isobornyl methacrylate, exo-norbornyl methacrylate, endo-norbornyl methacrylate or a mixture thereof in amounts within the above ranges, is essential in this invention to provide the present products having excellent solubility in ethanol/hydrocarbon propellant mixture and excellent hair holding, as shown in Examples 11-12.

EXAMPLE 14 (COMPARATIVE)

As discussed above, it is essential in the preparation of the polymers of the present invention by a suspension polymerization method that their molecular weights are sufficiently high to provide superior curl retention but insufficient to cause nozzle clogging in the dispenser. The following experiment was carried out to establish that the solution polymerization in ethanol failed to provide the adequate molecular weight of this invention.

To a one-liter glass resin kettle was added a reactant mixture of 68.80 g. of vinyl acetate (0.8 mole), 103.20 g. of mono-n-butyl maleate (0.6 mole), 16.64 g. of isobornyl acrylate (0.08 mole), 188.64 g. of anhydrous ethanol and 2.52 g. of di-(2-ethylhexyl)peroxydicarbonate (Lupersol 223 M75 75% active) and the mixture was stirred at 60°C. for 16 hours. The final product (Polymer 12) in ethanol was recovered

and had a solids content of 48.90%, a relative viscosity of 1.07 and a K value of 15. The curl retention of Polymer 12 was inferior to the same polymer (i.e. Polymer 7) having a K value of 48.5, prepared by the suspension polymerization method. A comparison of curl retention capability is summarized below in Table V.

TABLE V

Hair Spray Resin	Curl Retention Relative to Resyn® 28 - 2930*		
	K Value	60 Minutes	90 Minutes
Resyn® 28 - 2930 (control)	40	1.00	1.00
Polymer 7	48.5	1.23	1.13
Polymer 12	15	0.65	0.69

* Differences of 0.10 or greater represent significant difference in hair holding performance.

EXAMPLE 15 (COMPARATIVE)

The following example compares the clear/cloud temperatures, relative viscosities and spray patterns of the present resins with the commercial hair spray resin, Resyn® 28-2930. In all cases, the clear/cloud temperature was determined on 2.5% resin in anhydrous ethanol and the relative viscosities were determined on 1.0% resin in anhydrous ethanol. For spray pattern determinations, each composition was formulated with 2.5 parts of resin (100% neutralized with 2-amino-2-methyl-1-propanol), 67.5 parts of anhydrous ethanol and 30 parts of hydrocarbon propellant A-48 (blend of 20/80 propane/isobutane). The spray patterns from aerosol bottles fitted with a Misttelle I (Calmar) valve were recorded on alcohol-sensitive paper (Permapaper from Hewlett Packard Corporation) and results reported in following Table VI.

TABLE VI

Hair Spray Resin	Relative Viscosity (K value)	Clear/Cloud Temp., °F.	Spray Pattern
Polymer 1	1.66 (49.6)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 2	1.73 (51.8)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 3	1.60 (47.3)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 4	1.72 (51.3)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 5	1.41 (39.5)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 6	1.53 (44.6)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 7	1.64 (48.5)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 8	1.34 (35.6)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 9	1.61 (47.8)	below - 17.8° C (0° F.), both	Fine, disperse spray
Polymer 10	1.63 (48.4)	below - 17.8° C (0° F.), both	Fine, disperse spray
Resyn® 28 - 2930	1.21 (28.0)	- 7.8/ - 5° C (18/23° F.)	Fine, disperse spray

The hair spray resins of the present invention deliver the desired fine, disperse spray pattern without nozzle clogging and generally have a viscosity higher than the commercial hair spray resin Resyn® 28-2930. Furthermore, their solubilities are superior to the above commercial hair spray resin as indicated by their low clear/cloud temperatures. Also, no corrosion of the spray container was noted for the polymers of the present invention.

TABLE 16 (COMPARATIVE)

This example illustrates that the resin film of this invention (Polymer 7) is non-tacky under relatively high humidity conditions and is superior to the commercial hair spray resin Resyn® 28-2930.

Each hair spray resin was dissolved in ethanol (5%) and neutralized with 2-amino-2-methyl-1-propanol (AMP) to a level commonly found in commercial hair spray formulations - namely, Polymer 7,

10% and Resyn 28-2930, 90%. Each resin solution was then sprayed evenly (18 shots) on glass plates, using aerosol bottles fitted with a Misttelle I (Calmar) valve. After drying in air, three cotton balls were gently placed on each test area, and the plates transferred to a preconditioned humidity cabinet (temperature at 21.1°C (70° and 26.7°C (80°F.) and % relative humidity at 80% and 90%). At 20 minute intervals, one of the three cotton ball was pressed onto the film and then pulled away. The film tackiness was determined from the amount of cotton fiber retained by the film and was rated as NT (no tack), ST (slight tack), or T (significant tack). The results of these experiments are reported in Table VII.

TABLE VII

TESTING CONDITIONS		POLYMER 7	RESYN® 28-2930
21.1°C (70°F.) / 80% RH			
20 min.		NT	NT
40 min.		NT	NT
60 min.		NT	NT
26.7°C (80°F.) / 90% RH			
20 min.		NT	T
40 min.		NT	T
60 min.		NT	T

EXAMPLE 17

This example demonstrates that the hair spray resins of the present invention exhibit excellent water compatibility and thus significantly reduced flammability in both pump and aerosol hair-spray applications.

Hair spray resin, Polymer 7, (3.0 parts) was dissolved in ethanol, (SDA 40-2), (15 parts), neutralized with 2-amino-2-methyl-1-propanol, (0.50 parts) and blended with 81.5 parts of distilled water.

The clear solution thus obtained contains a 50% neutralized resin and is suitable for pump spray applications. When sprayed on the dry, clean tresses, the clear resin solution produced a non-tacky, clear, transparent film.

All of the products of this invention are non-corrosive to metal containers and their viscosities remain unchanged over long periods such that nozzle plugging in the normal life of an aerosol dispenser is avoided. It is to be understood that other species of vinyl ester and maleate half esters mentioned in the foregoing disclosure and other bicyclic esters or mixtures thereof including the exo- and endo-norbornyl acrylates or methacrylates or any intermixture of the present bicyclic esters can be substituted in the above examples to provide resin beads having improved solubility, propellant compatibility and curl retention.

Claims

1. A terpolymer of monomers (a) a vinyl ester; (b) a water insoluble or water miscible alkyl maleate half ester and (c) an acrylate or methacrylate ester of a saturated, hydroxylated, bicyclic hydrocarbon in a molar ratio of (a) to (b) to (c) of about 1:0.35 - 1:0.05 - 0.25.
2. The terpolymer of claim 1 wherein said vinyl ester is vinyl acetate and said acrylate or methacrylate monomer is isobornyl acrylate, isobornyl methacrylate or a mixture thereof.
3. The terpolymer of Claim 2 wherein said alkyl maleate half ester is mono-n-butyl maleate, mono-n-pentyl maleate or a mixture thereof.
4. The terpolymer of Claim 1 wherein said terpolymer is vinyl acetate/mono-n-butyl maleate/ an isobornyl acrylate in a molar ratio of 1:0.6 - 0.8:0.08 - 0.12.

5. Use of the terpolymer of any one of claims 1 to 4 in an inert carrier formulation containing an alcohol as a hair spray.
6. Use according to claim 5 wherein the carrier is an inert aqueous ethanol carrier.
7. Use according to claim 5 wherein the formulation contains a mixture of an alcohol and a hydrocarbon carrier.
8. Use according to claim 7 wherein said carrier is a mixture of ethanol, propane and isobutane and said formulation contains a propellant suitable for use in a hair spray.
9. Use according to any one of claims 5 to 8 wherein the terpolymer is employed at a concentration of between 0.5% and 10% solids in the formulation.
10. The process for producing the terpolymeric resin of Claim 1 as discrete microspherical particles which comprises: mixing monomers (a), (b) and (c) in the recited proportions with a low temperature free radical initiator; polymerizing said monomers in an inert suspension medium with constant agitation in a reaction zone maintained at a temperature between 40°C. and 90°C. until a microspherical particulate product is formed; separating said particles from the suspension medium and recovering said particles as the product of the process.
11. The process for producing the terpolymeric resin of claim 3 as discrete microspherical particles which comprises: mixing monomers (a), (b) and (c) in the recited proportions with a low temperature free radical initiator; polymerizing said monomers in an inert suspension medium with constant agitation in a reaction zone maintained at a temperature between 50°C. and 70°C. until the microspherical particulate product is formed; separating said particles from the suspension medium and recovering said particles as the product of the process.
12. The process of Claim 10 wherein said initiator is selected from the group consisting of a peroxyester, a peroxydicarbonate and an azo compound.
13. The process of Claim 11 wherein said initiator is selected from the group consisting of tertiary-butyl peroxyneodecanoate, tertiary-butyl peroxyneodecanoate, tertiary-butyl peroxyneodecanoate, tertiary-butyl peroxyneodecanoate, di-(2-ethylhexyl) peroxydicarbonate, di-(sec-butyl) peroxydicarbonate, di-(n-propyl) peroxydicarbonate, 2,2'-azobis-(2,4-dimethyl-4-methoxy-valeronitrile), and 2,2'-azobis-(2,4-dimethyl-valeronitrile).
14. The process of Claim 11 wherein said initiator is tertiary-butyl peroxyneodecanoate or tertiary-butyl peroxyneodecanoate.
15. The process of Claim 11 wherein said initiator is di-(2-ethylhexyl) peroxydicarbonate.
16. The process of Claim 10 wherein said initiator is employed in a concentration of between 0.05 and 5.0 wt. % based on the total monomers of the terpolymer.
17. The process of Claim 11 wherein said suspension medium is an aqueous solution of a copolymer selected from the group consisting of methyl vinyl ether/maleic acid copolymer and ethylene/maleic anhydride copolymer.
18. The process of Claim 17 wherein the aqueous suspension medium contains between 0.1 wt.% and 2.0 wt.% of said copolymer and the total monomers of the terpolymer are in a concentration between 10 wt.% and 50 wt.%.

Patentansprüche

1. Terpolymer aus den Monomeren: (a) einem Vinylester; (b) einem wasserunlöslichen oder wasser-mischbaren Alkylmaleathalbesten und (c) einem Acrylat- oder Methacrylatester eines gesättigten, hydroxylierten, bicyklischen Kohlenwasserstoffs in einem Molverhältnis von (a) zu (b) zu (c) von etwa

1:0,35 – 1:0,05 – 0,25.

2. Terpolymer nach Anspruch 1, worin der genannte Vinylester Vinylacetat ist und das genannte Acrylat – oder Methacrylatmonomer Isobornylacrylat, Isobornylmethacrylat oder eine Mischung davon ist.
- 5 3. Terpolymer nach Anspruch 2, worin der genannte Alkylmaleathalbesten Mono – n – butylmaleat, Mono – n – pentylmaleat oder eine Mischung davon ist.
4. Terpolymer nach Anspruch 1, worin das genannte Terpolymer Vinylacetat/Mono – n – butylmaleat/ ein Isobornylacrylat in einem Molverhältnis von 1:0,6 – 0,8:0,08 – 0,12 ist.
- 10 5. Verwendung des Terpolymers nach einem der Ansprüche 1 bis 4 in einer inerten Trägerformulierung, die einen Alkohol enthält, als ein Haarspray.
- 15 6. Verwendung nach Anspruch 5, worin der Träger ein inerte wässriger Äthanolträger ist.
7. Verwendung nach Anspruch 5, worin die Formulierung eine Mischung aus einem Alkohol und einem Kohlenwasserstoffträger enthält.
- 20 8. Verwendung nach Anspruch 7, worin der genannte Träger eine Mischung aus Äthanol, Propan und Isobutan ist und die genannte Formulierung ein Treibmittel enthält, das zur Verwendung in einem Haarspray geeignet ist.
9. Verwendung nach einem der Ansprüche 5 bis 8, worin das Terpolymer in einer Konzentration zwischen 0,5% und 10% an Feststoffen in der Formulierung eingesetzt wird.
- 25 10. Verfahren zur Herstellung des terpolymeren Harzes nach Anspruch 1 als diskrete mikrosphärische Teilchen, welches umfaßt: das Mischen der Monomere (a), (b) und (c) in den genannten Anteilen mit einem Niedertemperatur – Freiradikalinitiator; das Polymerisieren der genannten Monomere in einem inerten Suspensionsmedium unter ständigem Rühren in einer Reaktionszone, die bei einer Temperatur zwischen 40 °C und 90 °C gehalten wird, bis ein Produkt aus mikrosphärischen Teilchen gebildet ist; das Abtrennen der genannten Teilchen vom Suspensionsmedium und das Gewinnen der genannten Teilchen als das Produkt des Verfahrens.
- 30 11. Verfahren zur Herstellung des terpolymeren Harzes nach Anspruch 3 als diskrete mikrosphärische Teilchen, welches umfaßt: das Mischen der Monomere (a), (b) und (c) in den angeführten Anteilen mit einem Niedertemperatur – Freiradikalinitiator; das Polymerisieren der genannten Monomeren in einem inerten Suspensionsmedium unter ständigem Rühren in einer bei einer Temperatur zwischen 50 °C und 70 °C gehaltenen Reaktionszone, bis das Produkt aus mikrosphärischen Teilchen gebildet ist; das Abtrennen der genannten Teilchen vom Suspensionsmedium und das Gewinnen der genannten Teilchen als das Produkt des Verfahrens.
- 35 12. Verfahren nach Anspruch 10, worin der genannte Initiator aus der Gruppe ausgewählt ist, die aus einem Peroxyester, einem Peroxydicarbonat und einer Azoverbindung besteht.
- 40 13. Verfahren nach Anspruch 11, worin der genannte Initiator aus der Gruppe ausgewählt ist, die aus tert. – Butylperoxy-pivalat, tert. – Amylperoxy-pivalat, tert. – Butylperoxyneodecanoat, tert. – Amylperoxyneodecanoat, di – (2 – Äthylhexyl)peroxydicarbonat, di – (sec. – Butyl)peroxydicarbonat, di – (n – Propyl) – peroxydicarbonat, 2,2' – Azobis – (2,4 – dimethyl – 4 – methoxyvaleronitril) und 2,2' – Azobis – (2,4 – di – methylvaleronitril) besteht.
- 45 14. Verfahren nach Anspruch 11, worin der genannte Initiator tert. – Butylperoxyneodecanoat oder tert. – Butylperoxy-pivalat ist.
- 50 15. Verfahren nach Anspruch 11, worin der genannte Initiator di – (2 – Äthylhexyl)peroxydicarbonat ist.
- 55 16. Verfahren nach Anspruch 10, worin der genannte Initiator in einer Konzentration von zwischen 0,05 und 5,0 Gew.%, bezogen auf die gesamten Monomeren des Terpolymers, eingesetzt wird.

17. Verfahren nach Anspruch 11, worin das genannte Suspensionsmedium eine wässrige Lösung eines Copolymers ist, das aus der Gruppe ausgewählt ist, die aus Methylvinyläther/Maleinsäure - Copolymer und Äthylen/Maleinsäureanhydrid - Copolymer besteht.

5 18. Verfahren nach Anspruch 17, worin das wässrige Suspensionsmedium zwischen 0,1 Gew. - % und 2,0 Gew. - % des genannten Copolymers enthält und die gesamten Monomeren des Terpolymers in einer Konzentration zwischen 10 Gew. - % und 50 Gew. - % vorliegen.

Revendications

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1. Terpolymère de monomères comprenant (a) un ester vinylique; (b) un semi-ester maléique d'alkyle insoluble dans l'eau ou miscible avec l'eau et (c) un ester méthacrylique ou acrylique d'un hydrocar- bure bicyclicque hydroxylé, saturé, dans un rapport molaire de (a):(b):(c) d'environ 1;0,35 - 1:0,05 - 0,25.

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2. Terpolymère selon la revendication 1, dans lequel ledit ester vinylique est l'acétate de vinyle et ledit monomère acrylique ou méthacrylique est l'acrylate d'isobornyle, le méthacrylate d'isobornyle ou un mélange de ces substances.

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3. Terpolymère selon la revendication 2 dans lequel ledit semi-ester maléique d'alkyle est le maléate de mono - n - butyle, le maléate de mono - n - pentyle ou un de leurs mélanges.

4. Terpolymère selon la revendication 1, ledit terpolymère comprenant de l'acétate de vinyle/maléate de mono - n - butyle/un acrylate d'isobornyle dans un rapport molaire de 1:0,6 - 0,8 : 0,08 - 0,12

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5. Utilisation du terpolymère selon l'une des revendications 1 à 4 dans une formulation de véhicule inerte contenant un alcool comme produit de pulvérisation capillaire.

6. Utilisation selon la revendication 5, dans laquelle le véhicule est un véhicule d'éthanol aqueux inerte.

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7. Utilisation selon la revendication 5, dans laquelle la formulation contient un mélange d'un alcool et d'un véhicule hydrocarboné.

8. Utilisation selon la revendication 7, dans laquelle ledit véhicule est un mélange d'éthanol, de propane et d'isobutane et ladite formulation contient un propulseur convenant pour l'utilisation dans une lotion capillaire pulvérisable.

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9. Utilisation selon l'une des revendications 6 à 8, dans laquelle le terpolymère est utilisé à une concentration de 0,5 à 10% de substances solides dans la formulation.

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10. Procédé pour produire la résine terpolymère de la revendication 1 sous forme de particules micros- phériques discrètes, comportant les étapes de: mélanger les monomères (a), (b) et (c) dans les proportions citées avec un initiateur de radicaux libres à basse température; polymériser lesdits monomères dans un milieu de suspension inerte sous agitation constante dans une zone de réaction maintenue à une température de 40 à 90 °C jusqu'à ce qu'il se forme un produit en particules microsphériques; séparer lesdites particules du milieu de suspension et reprendre lesdites particules comme produit du procédé.

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11. Le procédé pour produire la résine terpolymère de la revendication 3 sous forme de particules discrètes microsphériques, comportant les étapes de: mélanger les monomères (a), (b) et (c) dans les proportions citées avec un initiateur de radicaux libres à basse température; polymériser lesdits monomères dans un milieu de suspension inerte sous agitation constante dans une zone de réaction maintenue à une température de 50 à 70 °C jusqu'à ce que le produit en particules microsphériques soit formé; séparer lesdites particules du milieu de suspension et reprendre lesdites particules comme produit du procédé.

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12. Procédé selon la revendication 10, dans lequel ledit initiateur est choisi dans le groupe comportant un peroxyester, un peroxydicarbonate et un composé azo.

13. Procédé selon la revendication 11 dans lequel ledit initiateur est choisi dans le groupe comportant le peroxyphthalate de butyle tertiaire, le peroxyphthalate d'amyle tertiaire, le peroxyneodécanoate de butyle tertiaire, le peroxyneodécanoate d'amyle tertiaire, le peroxydicarbonate de di-(2-éthylhexyl), le peroxydicarbonate de di(butyle secondaire), le peroxydicarbonate de di(n-propyle), le 2,2'-azobis-(2,4-diméthyl-4-méthoxy-valéronitrile), et le 2,2'-azobis-(2,4-diméthyl-valéronitrile).
14. Procédé selon la revendication 12, dans lequel ledit initiateur est le peroxyneodécanoate de butyle tertiaire ou le peroxyphthalate de butyle tertiaire.
15. Procédé selon la revendication 11, dans lequel ledit initiateur est le peroxydicarbonate de di-(2-éthylhexyle).
16. Procédé selon la revendication 10 dans lequel ledit initiateur est employé à une concentration de 0,05 à 5,0% en poids sur base des monomères totaux du terpolymère.
17. Procédé selon la revendication 11 dans lequel ledit milieu de suspension est une solution aqueuse d'un copolymère choisi dans le groupe comportant un copolymère d'éther méthylvinyle/acide maléique et un copolymère d'éthylène/anhydride maléique.
18. Procédé selon la revendication 17, dans lequel le milieu de suspension aqueuse contient de 0,1% en poids à 2,0% en poids dudit copolymère et le total des monomères du terpolymère représente une concentration de 10% en poids à 50% en poids.